METHYLMERCURY CYCLING AT THE SURFACE OF THE SEA: INTERACTION OF RAIN AND SEAWATER





MOTIVATION:

Methylated mercury can efficiently bioaccumulate in marine organisms even when seawater concentrations are extremely low (pg/L). It is therefore critical to understand which factors can significantly influence its stability in surface ocean. While photodegradation has been the main process TAKE A PICTURE OF THIS **QR CODE TO ACCESS POSTER ONLINE**



reported to degrade MeHg in surface ocean (DiMento & Mason 2017), we suggest that there are additional mechanisms to be considered.

STUDY AIM:

Evaluate the interaction of rain with seawater in microcosm experiments.

OVERARCHING HYPOTHESIS: Wet deposition, i.e. rain, to the sea surface could cause changes in mercury (Hg) speciation through processes other than dilution.

EXPERIMENTAL DESIGN: Seawater and rain mixed at different proportions with final volume of 125 mL per **bottle (x3 replicates per treatment –see** cartoon below) and immediately acidified (hydrochloric acid final conc. 0.5% v/v). Samples were digested with sulfuric acid (final conc. 1% v/v) prior to analysis by direct ethylation using Tekran 2700 system (Munson et al. 2014).

Mixing of seawater with rain can result in an 80% decline of total methylmercury concentration through



processes other than dilution.

0.12 0.10 0.08 0.06 0.04 0.02 0.00 15

FIGURE 1: Declining mean (error bar = 1 standard deviations) as the proportion of rain increases. Max average decline of 81%. Find the color legend in Table 1. **Detection limit for** MeHg_{Tot} was 0.011 ng/L. **J** 0.12 <u>ව</u> 0.10 **0.08** 0.06 \geq 0.04 0.02 0.00

FIGURE 2: A 77% decline in MeHg_{Tot} concentration from before to after the rain. **Colors correspond to Experiments 2 and 3 –** see Table 1.

Prior to Rain After Rain Time of seawater sampling

*Mackenzie Blanusa is an undergraduate student at UConn

Volume of rain (mL)

Table 2: The % of initial MeHg_{Tot} concentration as proportion Table 1: Information about the experiments including characterization of of rain increases from 0 to 12.5% (i.e. 15 mL rain in 125 mL storm types and rain properties. Left - color legend relates to Figure 1. of seawater) for each of the 4 experiments – see Table 2. THg in Rain NO₃⁻ in Rain **Seawater collected** Rain Experiment **Storm Type** Expt. 1 Expt. 4 Expt. 2 Expt. 3 Volume of rain added **Before or After Rain** Date (ng/L)(µM) % of initial MeHg_{Tot} concentration to seawater (mL) 9/6/18 **Squall Line** Before 30.7 37.7 Expt.1 100 100 100 100 0 Expt.2 Before Hurricane 9/10/18 3.5 48 102 100 103 4.6 Remnants After Expt.3 26 10 17 63 48 Extratropical 19 25 15 63 59 9/17/18 1.7 Expt.4 7.2 Before Cyclone

References: B. DiMento and R.P. Mason (2017) Factors controlling the photochemical degradation of methylmercury in coastal and oceanic waters. Marine Chemistry 196; 116-125. Munson, K.M., Babi, D., Lamborg, C.H., 2014. Determination of monomethylmercury from seawater with ascorbic acid-assisted direct ethylation. Limnology and Oceanography: Methods 12 (1); 1-9.